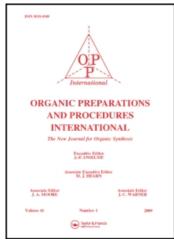
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A GENERAL AND CONVENIENT SYNTHESIS OF 2H-1,4-BENZOXAZIN-3(4H)-ONES

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OPPI BRIEFS

A GENERAL AND CONVENIENT

SYNTHESIS OF 2H-1,4-BENZOXAZIN-3(4H)-ONES[†]

Submitted by D. R. Shridhar*, M. Jogibhukta and V. S. H. Krish-(4/10/81) nan

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Previous syntheses of 2H-1,4-benzoxazin-3(4H)-ones (III) 1-3 suffer from one or more disadvantages such as low yields, long reaction times and contamination of the end product with by-products. The more recent method of Rufenacht et al.4, involving the treatment of o-aminophenol with chloroacetyl chloride in butanone in the presence of aqueous NaHCO3, when applied to the synthesis of the substituted benzoxazinones (IIIb-e), afforded the desired products in relatively low yields which were invariably contaminated with significant quantities of the corresponding uncyclized o-chloroacetylaminophenols.

$$\begin{array}{c} R_{2} \\ R_{1} \\ \end{array} \begin{array}{c} OH \\ NH_{2} \end{array} \begin{array}{c} + ClCH_{2}COC1 \\ \end{array} \begin{array}{c} 80-95\% \\ \end{array} \begin{array}{c} R_{2} \\ R_{1} \\ \end{array} \begin{array}{c} NH \\ H \\ \end{array} \begin{array}{c} OH \\ H \\ \end{array} \begin{array}{c$$

a)
$$R_1 = H_2 = H$$
 b) $R_1 = H$, $R_2 = NO_2$ c) $R_1 = C1$, $R_2 = H$

d)
$$R_1 = CH_3$$
, $R_2 = H$ e) $R_1 = NC_2$, $R_2 = H$

We now report that the use of isobutyl methyl ketone as solvent provides III in high yields and in excellent purity by

the reaction of the appropriate \underline{o} -aminophenol (I) with chloro-acetyl chloride in refluxing isobutyl methyl ketone in the presence of aqueous NaHCO3.

EXPERIMENTAL

General Procedure. To a solution of the appropriate o-aminophenol (1.0 mol) in isobutyl methyl ketone (600 ml), was added NaHCO₃ (200 g, 2.39 mol) and water (600 ml) and the resulting mixture was cooled in ice-bath. Chloroacetyl chloride (130 g, 1.15 mol) was added dropwise with stirring and the cold mixture was set aside to become ambient; it was then refluxed for 4 hrs and cooled to give the corresponding benzoxazinones in 80-95% yields. The products, thus prepared, had mps in close agreement with those reported in the literature as well as consistent IR ans PMR spectra.

- IIIa, 81% yield, mp. 172°, lit. 4 mp. 172°.
- IIIb, 92% yield, mp. 232°, lit. mp. 233°; pmr (DMSO-d₆): δ 4.66 (s, 2, CH₂), 7.0 (d, 1, J = 9Hz., C₅-H), 7.51 (d, 1, J = 3Hz., C₈-H), 7.83 (d/d, 1, J = 9 & 3Hz., C₆H) and 11.2 (s, 1, NH, exchangeable with D₂O).
- IIIc, 95% yield, mp. 214°, lit. mp. 215°; pmr (DMSO-d₆): δ 4.46 (s, 2, CH₂), 6.83 (s, 3, ArH) and 10.66 (s, 1, NH, exchangeable with D₂O).
- IIId, 85% yield, mp. 209° , lit. 2 209° ; pmr (DMSO-d₆): δ 2.13 (s, 3, CH₃), 4.4 (s, 2, CH₂), 6.53-6.83 (m, 3, ArH) and 10.83 (s, 1, NH, exchangeable with D₂O).
- IIIe, 87% yield, mp. 233°, lit. 1 mp. 233°.

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AN UNUSUAL SYNTHESIS OF 2,6-Di-t-BUTYL-4-ISOPROPENYLPHENOL

Submitted by Lajos Avar (5/22/81)

Sandoz Ltd. CH-4002 Basle SWITZERLAND

The reaction of 3,5-di- \underline{t} -butyl-4-hydroxy- α -bromo- α -methyl-propiophenone(I)^{1,2} with sodium hydroxide leads to 2,6-di- \underline{t} -butyl-4-isopropenylphenol (II), described earlier by Braun and Maier³ by dehydration of 2(3,5-di- \underline{t} -butyl-4-hydroxyphenyl)-2-propanol with neutral aluminium oxide.

HO CO-C-Br OH CH₃

$$CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

$$CH_{3}$$

EXPERIMENTAL

The IR spectra were measured with a Perkin-Elmer 257 Grating Infrared Spectrophotometer. The NMR spectra were determined with A Bruker Spectro Spin WP 60 with TMS as internal standard.